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# Phytochemical Screening, Antioxidant, Antiulcer, Anti-Inflammatory and Analgesic Activity of the Aqueous Extract of Angelica archangelica

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#### ARTICLE INFO

#### ABSTRACT

Research Article

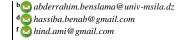
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The objective of this study is to evaluate the antioxidant, antiulcer, anti-inflammatory and analgesic activity of the aqueous extract of Angelica archangelica L., a medicinal plant of the traditional pharmacopoeia of Algeria. The aqueous extract showed a large amount of total polyphenols, tannins, chlorophyll a, b, carotenoids and a small amount of flavonoids with values of 80.16±14.3 mg EAG/g of extract, 3.12±2.01 mg EQ/g of extract, 226.10±4.50 mg EAT/g of extract, 6.10±50.62, 12.00±31.53 and 1.78±0.40 μg/mL respectively. In addition, the aqueous extract of A. archangelica showed a high ability to trap DPPH radical in with IC50 at 16.9±2.03μg/mL. However, the aqueous extract has a great protective effect against β-carotene degradation at 91.98±0.64% after 24 hours and significant ferrous ion chelation activity at 46.2 ±1.53 µg/mL. Furthermore, the antiinflammatory activity was studied according to the protein denaturation inhibition method and, according to the results obtained, the extract of A. archangelica at concentrations of 250, 500 μg/mL and 1 mg/mL have ahigh antiinflammatory activity whose inhibition percentages are 78.85±5.31%, 86.65±2.70% and 89.89±0.58% respectively. The analgesic effect of the aqueous extract of A. archangelica, showed that the concentration 400 mg/mL has a greet abdominal cramps inhibitory effect in comparison with the concentration 200 mg/mL with a percentage of 98.28%. The antiulcer effect of the aqueous extract of A. archangelica was evaluated by the 70% ethanol-induced ulcer test. The results obtained reveal that the aqueous extract 200 and 400 mg/mL exerted a considerable effect of protecting the stomach at 86.55±3.51% and 82.82±2.18% respectively.



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## Introduction

The action of eating and drinking provides our body with the elements necessary for maintenance of life and health, Food will be varied, sufficient, adapted to the needs of each individual, in correlation with his activity (Akgul et al., 2020; Kına et al., 2021). From food, the body makes raw material that will play the role of structural materials and energy source of our organization (Sevindik et al., 2017; Uysal et al., 2021). Nutritional status also depends on digestion by the digestive system which digests the food and absorbs the elements that it transforms in the blood or lymph by making them cross its mucous membrane and eliminate waste (Korkmaz et al., 2021; Pehlivan et al., 2021).

Any agent that breaks the continuity of the gelatinous barrier formed by the mucosa causes inflammation of the stomach lining; this condition is called gastritis a persistent damage to the underlying tissue can lead to peptic ulcers, or more specifically gastric ulcers (Sipponen and Maaroos, 2015). This condition affects about 10% of adults in countries Westerners with a prevalence of 2% and an incidence of 0.03%. Gastric ulcer was also common in men than in women. The average age of onset of ulcers is 65 (exceptional before 40 years) (Karila-Cohen et al., 2005). The aetiologies of gastric ulcer were multiple: factors hereditary in 15 to 50% with a predominance of blood groups A, syndromes genetic (multiple endocrine neoplasias

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type 1), psychological factors (stress, stay in intensive care), environmental factors (tobacco, alcohol, coffee) and medication (salicylates, NSAIDs and AIS) (Mustafa et al., 2015; Ommurugan and Rao, 2019). It is currently believed that *Helicobacter pylori* is present with a prevalence of more than 60% in patients with a gastric ulcer and more than 70% in those with antral gastritis. This prevalence also clearly increases with age (Karila-Cohen et al., 2005).

This disease, easy to contract and difficult to cure, has serious repercussions on the health of the individual and can even lead to death. Its treatment is sometimes difficult, according to the depth of the ulcer. Treatment with modern medicine is sometimes expensive, which constitutes a handicap for the patient (Barkun and Leontiadis, 2010; Ommurugan and Rao, 2019). For those who have the means to do it, they develop sometimes a reluctance towards the disease, which is added to the side effects of medications. In Algeria, several plants are used to treat diseases inflammatory, especially gastric ulcer (Bouasla and Bouasla, 2017; Bouzabata, 2017). Among these plants, *Angelica archangelica* L.

A. archangelica commonly known as Angelica is a biennial or perennial herbaceous medicinal plant, aromatic characteristics of the Apiaceae family. European, distributed in mid-mountain regions of Europe and Asia (Chauhan et al., 2016), this plant is also widespread in Algeria (Bouzabata, 2017). Preferred growing media are moist, nutritious soil which can be easily removed from the roots after harvest, and is not sensitive to temperatures below 0 °C. The normal life cycle of A. archangelica is completed in three years, but sometimes, when the seedlings are well developed, it can be achieved in just two years (Kylin, 2010).

This genus has been used for centuries in traditional medicine for the treatment of a variety of illnesses, including influenza, cough, and inflammation (Rajtar et al., 2017). Angelica (A. archangelica) has real antibacterial, antifungal, antioxidant, anti-inflammatory, antiviral, anticonvulsant, sedative, diuretic, immunological and vasodilating properties, It would also relieve certain disorders such as: anemia, insufficient appetite, respiratory problem (asthma), digestion (gastric acidity, gastric ulcer, bloating, aerophagia, vomiting, etc.), male sexual fatigue, hepatism (Lacoste, 2003). In this cntexte the objectives of our work are: the phytochemical study of the aqueous extract of A. archangelica and the evaluation of the antioxidant, anti-inflammatory and analgesic activities of the extract.

#### **Materials and Methods**

### Materials

The angelica plant (*A. archangelica*) used in this study was purchased from an herbalist from Setif, Algeria. The plant was powdered and stored until use.

Albino-type mice, averaging 25g, were used to study the biological activity of the aqueous extract *in vivo*. The latter come from the breeding centers of the Pasteur Institute in Algiers. The experiment was carried out at the animal facility of the University of Setif 1. The mice were raised in conditions favorable to their growth and development, with a light cycle of 12 hours per day and free access to food and water.

All the products used in this study come from Sigma (Paris, France). Carbo methyl cellulose (CMC 1.5%), acetic acid (CH<sub>3</sub>COH), Methanol, ethanol, chloroform, linoleic acid (C18H30O2), gallic acid and tannic acid. The chemical reagents are: 2,2'-diphenyl-1-picryl hydrazyl (DPPH), trichloride aluminum (AlCl<sub>3</sub>), Folin – Ciocalteu, Sodium bicarbonate (Na<sub>2</sub>CO<sub>3</sub>), Sodium chloride (NaCl), Quercetin, Tween 40,  $\beta$  Carotene (C<sub>40</sub>H<sub>56</sub>) and Butylhydroxytoluene (BHT).

#### Methods

The angelica plant was crushed using a grinder to obtain a powder. A quantity of 30 g of this powder was boiled in 1 L of water for 10 minutes and allowed to cool. When the decoction period was over, the homogenate obtained was filtered through the muslin, and vigorously squeezing everything in order to extract as much liquid as possible. The filtrate obtained was subjected to further filtration through filter paper, and poured into plates, then dried in an oven at 38°C (Perera et al., 2008). The extract obtained was then stored in a cold place at 4°C in a protected bottle, in order to avoid any degradation of the molecules by light.

The extraction yield (%) (extractable components) expressed on dry weight basis of pant materiel was calculated from the following equation: Yield  $(g/100 \text{ g}) = (W1\times100)/W2$  where W1 is the weight of the extract residue obtained after solvent evaporation and W2 is the weight of pant materiel used in the extraction.

#### Phytochemical Analysis

Total polyphenols

The determination of the total polyphenols was carried out with the colorimetric reagent Folin Ciocalteu. This reagent consists of a mixture of phosphotungstic acid  $(H_3PW_{12}O_{40})$  and phosphomolybdic acid  $(H_3PM_{12}O_{40})$ . This reagent is reduced upon oxidation of phenols to a mixture of greenish blue oxide of tungsten and molybdenum which has maximum absorption at 765 nm (Li et al., 2007). 500 µl of Folin-Ciocalteu reagent (diluted 10 times in distilled water) are added to 100 µL of extract or standard solution of gallic acid. After 4 min, 400 µL of sodium carbonate solution (Na<sub>2</sub>CO<sub>3</sub>) with a concentration of 75 mg/mL are added to the reaction mixture. After an incubation of 1 hour 30 minutes at room temperature and in the dark, the absorbance is read at 765 nm using a spectrophotometer. The concentration of total polyphenols in the methanolic extract is calculated from the regression equation of the calibration range established with gallic acid and the results were expressed in micrograms of equivalent of gallic acid per milligram of dry extract (µg EAG / mg E).

Flavonoids

Aluminum trichloride (AlCl<sub>3</sub>) method was used to quantify flavonoids in the extract to be assayed (Benslama and Harrar, 2016). A volume of 500  $\mu L$  of the extract diluted in water and of different concentration of quercetin in methanol was added to 500  $\mu L$  of the AlCl<sub>3</sub> solution (2%). After 10 min of incubation at room temperature and in the dark, the absorbances are measured by the spectrophotometer at 430 nm. The concentration of flavonoids in the methanolic extract was calculated from the regression equation of the calibration range established with quercetin and expressed in micrograms of quercetin equivalent per milligram of extract ( $\mu g$  EQ / mg E).

**Tannins** 

To determine the tannins present in the extract, 1 mL of hemolyzed bovine blood (absorbance=1.6) was mixed with 1 mL of extract and left to stand for 10 min. The absorbance of the supernatant (4000 rpm for 10 min) was measured at 576 nm using a spectrophotometer against a white solution (Gharzouli et al., 2002). The calibration curve is constructed with tannic acid at different concentrations. The precipitation efficiency of the solutions tested is expressed in micrograms of tannic acid equivalent per milligram of extract (μg EAT / mg of extract).

Total chlorophylls and carotenoids

To determine the total chlorophylls and carotenoids in the extract, 1 mg of extract was dissolved in 1 mL of distilled water (Lichtenthaler and Wellburn, 1983). The absorbance was measured by a spectrophotometer at different wavelengths (450 nm, 645nm, 663nm). The results were expressed in micrograms per milliliter  $(\mu g/mL)$  using the following formulas:

 $\begin{array}{ll} \hbox{Chlorophyll a} &= 12.7 \times A \ 663\text{-}2.69 \times A \ 645. \\ \hbox{Chlorophyll b} &= 22.9 \times A \ 645\text{-}4.68 \times A \ 663. \\ \hbox{Total chlorophyll} &= 20.2 \times A \ 645 + 8.02 \times A \ 663. \\ \hbox{T carotenoids} &= 4.07 \times A \ 450 \ - \ [(0.0435 \times \text{Chlorophyll b})]. \\ \hbox{a)} &+ (0.367 \times \text{Chlorophyll b})]. \end{array}$ 

#### The Antioxidant Capacity

DPPH radical scavenging

The antioxidant capacity of the extract was estimated by using the DPPH radical scavenging test according to Benslama et al. (2017). In this test, the antioxidants reduce the diphenyl picryl hydrazyl (DPPH) having a purple color to a yellow compound, the color intensity of which of DPPH is inversely proportional to the capacity of the antioxidants present in the medium. The assessment of antioxidant capacity is achieved as follows. 1250  $\mu L$  of a methanolic solution of DPPH (0.04%) was mixed with 50  $\mu L$  of the plant extract or standard antioxidants (acid gallic) at different concentrations. The resulting mixture was then incubated away from the light at room temperature for 30 minutes. Absorbance was measured at 517 nm (Shekhar and Anju, 2014). The percentage inhibition (I%) is calculated according to the next formula:

RSA (%) = 
$$[(A_C - A_S) / A_C] \times 100$$

where RSA is Radical-scavenging activity,  $A_C$  is the absorbance of control and  $A_S$  is the absorbance of sample. The data were presented as half maximal effective concentrations (IC<sub>50</sub>), i.e. the concentration of the extract that is necessary to scavenge 50% of the DPPH radicals.

Beta-carotene (β-carotene) bleaching test

The oxidation of linoleic acid generates peroxide radicals, these free radicals will subsequently oxidize  $\beta$ -carotene causing the disappearance of its red color, which is followed spectrophotometrically at 490 nm. However, the presence of an antioxidant could neutralize free radicals derived from linoleic acid and therefore prevent the oxidation and bleaching of  $\beta$  carotene (Kartal et al., 2007). This test was carried out as follows: 0.7 mg of  $\beta$ -carotene were dissolved in 1 mL of chloroform. The resulting solution was placed in a flask containing 25µL of linoleic acid and 200 mg of Tween 40. After evaporation of the

chloroform, 100 mL of distilled water saturated with oxygen ( $H_2O_2$ ) was added with vigorous stirring. From this new solution 2.5 mL are transferred into tubes and 350  $\mu$ L of the extract (2 mg / mL) and of the control BHT, methanol, distilled water was added. The absorbance was immediately measured at 490 nm. Other readings are measured at different time intervals (1h, 2h, 4h and 24h).

Chelation power on Ferrous (Fe2+) Ions

The chelating power of the aqueous extract of A. archangelica on ferrous (Fe<sup>2+</sup>) ions was carried according to (Decker and Welch, 1990). A volume of  $50\mu$ L of FeCl<sub>2</sub> (0.6 mM) is added to a volume of  $250~\mu$ L of the extract solution and  $450~\mu$ L of methanol. After stirring, a volume of  $50~\mu$ L of ferrozine is added to the mixture. Everything is incubated for 10~minutes at room temperature and in the dark. The absorbances are read using a spectrophotometer at 562~m. EDTA was used as a standard for chelation. The percentage of iron chelation was determined according to the following formula:

% chelation = 
$$(A_C - A_E) / A_C \times 100$$

where  $A_C$  is the absorbance of control and  $A_S$  is the absorbance with the extract. The data were presented as half maximal effective concentrations (IC<sub>50</sub>), i.e. the concentration of the extract that is necessary to cheat 50% of the DPPH radicals.

in vitro Anti-inflammatory activity

The *in vitro* anti-inflammatory activity of extract was determined using the method of denaturation of albumin by heat. A quantity of 500  $\mu L$  of the extract or at different concentrations were added to 500  $\mu L$  of the egg albumin 0.2% (in Tris-HCl 50 mM, pH 6.6). After stirring, the solutions obtained are placed in an oven at 28 °C for 15 min. then are incubated in a water bath for 10 min (70 °C). Turbidity was measured at 660 nm using a spectrophotometer against a blank (Padmanabhan and Jangle, 2012). The non-steroidal anti-inflammatory drug diclofenac was used as a standard anti-inflammatory agent.

## Analgesic Effect

Before each experiment, mice are placed in individual cages thoroughly wire mesh to avoid coprophagia, and are deprived of their food for 12 to 18 hours, but have plenty of water until one hour before the experiment.

The measurement of the analgesic effect was based on the oral administration of 120  $\mu$ L (200 mg/kg and 400 mg/kg) of the plant extract. One hour later, 120  $\mu$ L of acetic acid was gently injected intraperitoneally. Then the mouse was left to stand for 5 min. Afterwards, the mouse was monitored for 25 min and the number of abnormal contractions is calculated (Atta and Alkofahi, 1998).

## Anti-ulcer Activity

The anti-ulcer activity f extract was carried out described method of mucosal damage injury induced by oral administration of the agent ulcerogenic (ethanol) one hour after administration of extract solutions (Gharzouli et al., 1999). The aqueous extract of A. archangelica at doses of 200 and 400 mg/kg was administered orally (125  $\mu L$  / mouse) one hour before the force-feeding of 125  $\mu L$  of ethanol. Mice in the negative control group received 1.5%

CMC by the same route in addition to oral administration of ethanol but mice in the positive control received 5 mg/kg ranitidine in addition to oral administration of ethanol. 30 minutes later, the mice are sacrificed by cervical dislocation. After a ventromedial laparotomy, the stomach was excised, then opened according to the greater curvature and washed with water and spread on a plate to determine the surface of the lesions using software Image J 1.5.

# Statistical Analysis

The results of different experiments are expressed as mean  $\pm$  SD (standard deviation) for in vitro tests and mean  $\pm$  SEM (mean standard deviation) for *in vivo* tests. The different IC<sub>50</sub> values for the *in vitro* test are calculated using the Graph pad prism 7 software according to the log (concentration) vs Normalization (absorbance) method. To compare the different values, the analysis of variance (one way ANOVA) was performed followed by the Tukey test (multiple comparison). The difference is considered statistically significant at the risk of 5% (P<0.05).

### **Results and Discussion**

## Phytochemical Screening

The plant parts were extracted using the decoction method, and gives a yield of 20.86%. This yield decreases between 9.8% and 11.88% during extraction using 95% methanol at 9.8% (Kumar and Bhat, 2012; Kumar et al., 2012). This difference can be attributed to the method of the extraction (Perera et al., 2008).

The determination of total polyphenols by the Folinciocalteu method makes it possible to observe that there is a correlation between the absorbance and the concentration of gallic acid (Tabel 1). The polyphenol assay results show that the aqueous extract of the leaves of *A. archangelica* contains a total polyphenol content of 80.16±14.3 mg EAG/g of extract.

The quantitative determination of total flavonoids by the AlCl<sub>3</sub> method in using quercetin as a standard, shows that absorbance increases with increasing concentration (Tabel 1). The result of the assay indicates that the aqueous extract contains an amount of flavonoids of  $3.12\pm2.014$  mg EQ/g E. The quantification of the tannins present in the aqueous extract of *A. archangelica* using the hemoglobin precipitation method showed that the extract contains  $226.1\pm4.5$  mg EAT/g E (Tabel 1).

The quantification of chlorophylls a, b and carotenoid, present in the aqueous extract of the A. archangelica, using the method of Lichtenthaler (1983), showed that the extract contains total chlorophyll a, b and carotenoid of  $6.10\pm50.62$ ,  $12.00\pm31.53$ and  $1.78\pm0.40$ respectively. Different studies show that different extracts from different parts of the A. archangelica contains polyphenols and flavonoids. However, the aqueous extract of plant has the highest content (Kumar et al., 2011). For different types of pigments, our results show that the content of chlorophyll a, b and carotenoids in A. A. archangelica are higher compared to that of turkie (Lobiuc et al., 2012). The comparison of our data with the published literature is rather difficult because variations have observed for the total phenolic content that could be attributed to the different varieties Angelica's solvent, extraction temperature and technique thus employed different parts used (Kumar and Bhat, 2012; Kumar et al., 2011). However the variation in different pigment contents may result from the influence of the harvest season which is affected by the period of fluoridation and fruit formation (Lobiuc et al., 2012).

#### **Antioxidant Activity**

The anti-free radical activity was carried out using DPPH method. 1-picrylhydrazyl (DPPH) which is a frequently used method for its simplicity. This method was based on the reduction of a methanolic solution of DPPH in the presence of an antioxidant which gives a hydrogen or an electron, the non-radical form DPPH-H was formed (Benslama et al., 2021, 2016). The inhibition of DPPH radical discoloration is a function of the concentration of the extract used and the control gallic acid (reference antioxidant). The antioxidant activity of the extract was expressed as IC50. The aqueous extract of *A. archangelica* presents an IC50 of  $16.9\pm2.03~\mu g/mL$  and  $1.07\pm0.007~\mu g$  mL for the standard (Figure 1).

The chelating activity of ferrous iron was measured by inhibition of formation of Fe2+-ferrozine complex after incubation of A. archangelica extract with Fe2+ according to the method of Decker and Welch (1990). Ferrozine can quantitatively form complexes with Fe<sup>2+</sup>. However, in the presence of chelating agents, the formation of the complex is disturbed so that the color of the complex is diminished. The measurement of the reduction in color, therefore, allows the estimation of the activity of chelation of the coexisting chelating agent. The divalent ferrous ion plays an important role as catalysts of oxidative processes, leading to the formation of superoxide radicals and hydroxyl anions by Fenton reactions. Chelation of ferrous ions (Fe<sup>2+</sup>) can make Significant antioxidant effects by delaying metal catalyzed oxidation. In this test our extract shows a chelation activity of 46.2±1.53 µg/mL (Figure 2).

The  $\beta$ -carotene bleaching mechanism is a radical mediated phenomenon free resulting from hydroperoxides formed from linoleic acid, these free radicals will subsequently oxidize the  $\beta$ -carotene, thereby causing the disappearance of its red color, which is followed spectrophotometrically at 490 nm. However, the presence of an antioxidant could neutralize radicals derived from linoleic acid and / or inhibit oxidation thus preventing bleaching of  $\beta$ -carotene. The kinetics of the activity of the aqueous extract of *A. archangelica* and BHT, methanol, water for a time interval between 1 hour to 24 hours presented in the figure (fig. 20). According to the results, a very significant anti-lipid peroxidation activity (at 24 h) was obtained with the aqueous extract and BHT 91.98±0.64%, 99.52±5.82% respectively.

Table 1. Phytochemical screening of the aqueous extract of *A. archangelica*.

Properties	Value
Yield	20.86%
TPC (µg EGA/mg E)	$80.16 \pm 14.30$
TFC (µg EQ/mg E)	3.12±2.01
Tannins	226.10±4.50
Chlorophylls µg/mL	a: 6.10±50.62 b: 12.00±31.53
Carotenoid µg/mL	1.78±0.40

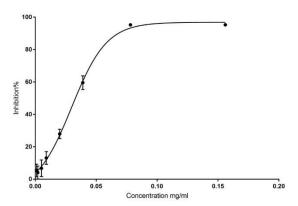


Figure 1. DPPH bleaching inhibition of *Angelica* archangelica aqueous extract

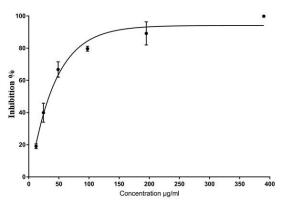


Figure 2. Ferrous ion chelating activity of *A. archangelica* aqueous extract

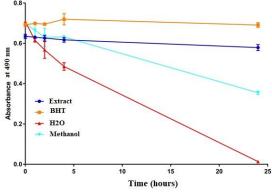


Figure 3. B carotene anti-bleaching activity of *A. archangelica* aqueous extract

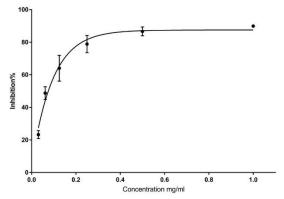


Figure 4. Albumine denaturation inhibitory effect of *A. archangelica* aqueous extract

Comparing our data with published literature is rather difficult because studies on the antioxidant activity of A. archangelica are minor almost unobtainable, however several, several studies showing the antioxidant effect of several varieties of the A. archangelica. The aqueous and ethanolic extracts of A. archangelica presents an antioxidant activity close to 80 and 20% respectively (100  $\mu g/mg$ ) (Heo and Lee, 2008). The activity protective against bleaching of  $\beta$ -carotene may be a result of the activity of plant compounds such as polyphenols and coumarins (Aminjafari et al., 2016).

# Anti-Inflammatory Activity

Analgesic effect

In mice, the pain is manifested as abdominal cramps. From the results, 0.6% acetic acid was found to induce an average of 66.33 cramps counted after 25 minutes in the control lot. The aqueous extract of A. archangelica at a concentration of 200 mg/kg, 400 mg/kg and sodium Diclofenac 10 mg / kg reduce (P<0.05) the number of abdominal cramps; the percentages of inhibition were: 53.62%, 98.28% and 58.43% respectively (Figure 5). According to previous studies, analgesic activity extracts may be due to the presence of saponins, phenolic compounds or alkaloids (Kumar et al., 2011). The analgesic effect of these compounds is confirmed in several medicinal plants products like Aloe vera, Mentha peperita, Eucalyptus camaldulentis, Jesminum officinalus (Atta and Alkofahi, 1998). These components may have the ability to inhibit cyclooxygenesis -1 and 5- lipooxygenase and / or inhibition of nerve transmission by the reabsorption blockade (Härmälä et al., 1992; Roos et al., 1997).

Anti-ulcer activity

The anti-ulcer effect of the aqueous extract of *A. archangelica* was determined. However, the oral administration of 70% ethanol to mice causes bleeding lesions extensive in the glandular part of the stomach. The formation of gastric lesions induced by ethanol, mice are orally administered the extract one hour before administration of the ulcerogenic agent at the dose tested (200 and 400 mg/kg) (fig. 23). 70% ethanol induces lesions on average 20.45±3.17%. Pretreatment of mice by the aqueous extract 200, 400 and ranitidine 5 mg/kg decreases the percentage ulceration. This decrease was expressed as a very close protection percentage (P>0.05) 86.55±3.51%, 82.82±2.18% and 84.62±4.56% for the aqueous extract 200, 400 and ranitidine 5 mg/kg respectively (Figure 6).

Studies on the protective effect of alcoholic root extract of A. archangelica with different doses (5 and 10 mL/kg) induces a protection of 30 and 80%. This protection is due to decreased acid secretion and decreased pepsin activity and quantity leukotriene present in gastric juice even by the increase in the concentration of mucins and prostaglandins (Khayyal et al., 2001). Other study shows that polysaccharides extracted from Angelica sinensis induces a decrease in the surface area percentage of gastric ulcer induced by acetic acid and inhibits angiogenesis protects the stomach developed cancer (Ye et al., 2003). The mechanism of the onset of pain results from tissue damage responsible for increased release of many chemicals mediators such as histamine, prostaglandin, and serotonin, in the intravenous fluid peritoneal, which will stimulate the nociceptive receptors located at the peritoneal level.

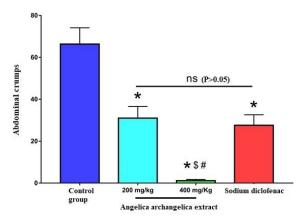


Figure 5. Analgesic effect of *A. archangelica* aqueous extract

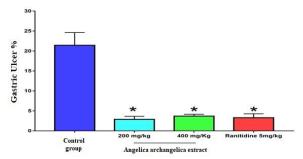


Figure 6. Anti-ulcer activity of *A. archangelica* aqueous extract

### Conclusion

Aromatic and medicinal plants are an inexhaustible source of substances having a wide variety of biological and pharmacological activities. As part of this study, we were interested in the extraction, dosage, the study of the analgesic, antiulcer, anti-inflammatory and anti-oxidant properties of the aqueous extract of *A. archangelica*. These properties are probably related to the presence of polyphenols, flavonoids and tannins demonstrated by the phytochemical study carried out.

The antioxidant activity of the extracts has been evaluated by numerous *in vitro* tests; the results obtained confirm that the extract is endowed with a strong antioxidant power. The results obtained by the analgesic study showed that the extract has a significant analgesic activity.

In the anti-inflammatory activity, the results have shown that A. archangelica has a strong inhibitory effect on the denaturation of proteins causing inflammation. During this study we demonstrated the protective effect in vivo of the aqueous extract of A. archangelica against gastric ulcer induced by the ethanol. The treatment with the extract helps to provide protection against ulcer, which means that the extract has potent anti-ulcer activity. The extract of A. archangelica would therefore constitute an advantageous source of medicine. traditional improved very accessible and would be cheaper for the populations. For that he it is better to study other effects of this plant such as anti-inflammatory effect in vivo and to determine which are the pure molecules responsible for these biological effects.

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